

PHY 101 Fall 2006
Formula Sheet for Final Exam
Stolkin Auditorium, Thurs Dec 14, 12:30 p.m.

$$KE = \frac{1}{2}mv^2.$$

$$GPE = mgh.$$

$$Elast.PE = \frac{1}{2}kx^2.$$

$$W = F_x * \Delta x.$$

$$W_{total} = \Delta KE$$

$$W_{external} = \Delta PE$$

$$P = \frac{\Delta E}{\Delta t}.$$

$$F_{grav} = mg \text{ near the Earth's surface; } F_{grav} = G \frac{m_1 m_2}{r^2} \text{ in general, where } G = 6.67 \cdot 10^{-11}$$

$$\text{m}^3/\text{kg}\cdot\text{sec}^2$$

$$F_{spring} = kx.$$

$$F_{elect} = k_e \frac{q_1 q_2}{r^2}, \text{ where } k_e = 8.99 \cdot 10^9 \text{ N m}^2/\text{C}^2.$$

$$F_{elect} = qE.$$

$$Q = CV$$

$$P_{elect} = V * I.$$

$$V = I * R.$$

$$\Phi = NBA; \quad V_{induced} = \Delta\Phi/\Delta t$$

$$F_{mag} = B * I * L$$

$$Elect.PE = \frac{1}{2}CV^2.$$

Electrochemical Energy stored in battery = Voltage * Capacity

$$H + W = \Delta E.$$

$$H = C m \Delta T$$

$$E = m c^2.$$

$$pV = nRT$$

$$pV = Nk_B T$$

$$\text{Thermal Energy in monatomic gas} = (3/2) * N * k_B * T = (3/2) * n * R * T$$

$$\text{Thermal Energy in diatomic gas} = (5/2) * N * k_B * T = (5/2) * n * R * T$$

$$W = p\Delta V$$

$$R = 8.314 \text{ J/mole} - \text{deg; } k_B = 1.381 * 10^{-23} \text{ J/deg}$$

$$T = 2\pi \sqrt{\frac{m}{k}} = 1/f$$